



Article

Influence of Sweat on Joint and Sensor Reliability of E-Textiles

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Abstract: This article addresses reliability under the sweat of interconnection techniques for the mounting surface mounted device (SMD) components and fully printed humidity sensors onto conductive stretchable textile ribbons. Samples underwent testing for the effect of ageing by artificial sweat on their electrical resistance using both alkaline and acidic artificial sweat. The best results in terms of electrical resistance change were obtained for samples soldered to the conductive fibers interwoven in the ribbon. However, this method can damage the ribbon due to the high temperature during soldering and significantly reduce the mechanical properties and flexibility of the ribbon, which can lead to a limited service life of samples. On the other hand, adhesive bonding is a very interesting alternative, where the above-mentioned properties are preserved, but there is a significant effect of sweat ageing on electrical resistance. The results of fully printed graphene-based humidity sensors show that, for the intended use of these sensors (i.e., detection of changes in moisture on the human body), usage of the samples is possible, and the samples are sufficiently reliable in the case of sweat degradation. In addition, the response of the sensor to humidity is quite high: 98% at a relative humidity of 98%.

Keywords: e-textiles; conductive stretchable textile ribbon; nonconductive adhesive compression bonding; soldering onto textile; fully printed graphene-based humidity sensor



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1. Introduction

At present, the popularity of electronic textiles (e-textiles) as a class of smart textiles is growing according to the wide range of potential applications, such as in the healthcare, sports, fitness, fashion, telecommunications, automotive and aerospace industries [1,2]. A combination of textile-based circuits and conventional passive components, or fully printed passive sensors, seems to be a reasonable solution for designing the smart textiles used in many novel applications, such as safety elements, healthcare or well-being products, and sport garments [3-5]. One of the important questions is whether surface-mounted device (SMD) components can be attached to textile-based substrates with properties (especially reliability) comparable to those of standard rigid fiberglass-reinforced boards (PCBs) with attached components. The greatest difference between smart textiles and common wearable electronics is that the sweat from any activity is absorbed by the textile substrate, and this process can affect the electrical parameters. For hygienic reasons, the textile also sometimes needs to be washed, which could also affect the joint properties [6]. Moreover, consumers expect e-textiles to retain the features of normal clothes, such as wearing comfort, elasticity, washability, breathability, and the use of hypoallergic materials that are nonirritating for skin, while providing added value functions such as heating, active illumination, and environmental or body parameter monitoring. This is a major challenge as all these aspects must be taken into consideration in e-textile design [7–9].

One of the possible materials for smart textile use is electrically conductive textile ribbons. These ribbons are flexible and stretchable, and the electrically conductive contact of an SMD component can be made with a conductive track woven in the ribbon, which

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was proven in our previous experiments [10,11]. The disadvantage of these ribbons is a low resistance to higher temperatures. The standard connection method on rigid printed circuit boards (fiberglass-reinforced epoxy substrates) is soldering or, alternatively, using electrically conductive adhesives. As mentioned, for textile substrates, the higher temperature used for the standard soldering process can be a problem. However, low-temperature soldering, while maintaining an acceptable temperature and a short soldering process time, is possible [12,13]. Alternatively, adhesive bonding with an electrically conductive adhesive (ECA) can be employed. The disadvantage of this process is the long curing time, and another process step is necessary for encapsulation. Another possibility is adhesive bonding with a nonconductive adhesive (NCA) to fix the components onto the substrate, and a conductive joint made by the direct contact of a component with a substrate is possible [14]. The problem with thermosetting adhesives is their long curing times, but a UV-curable adhesive with almost instant curing can be used. For the use of soldering while maintaining a low temperature [15], an ECA [16] or an NCA seems to be the easiest and cheapest way for contacting nearly all types of SMD components to a textile while maintaining the wearing comfort and good reliability [17]. The soldering and ECA techniques create electrical contact between components and conductive threads via solder or conductive particles in the adhesive. The NCA contact technique creates direct contact between conductive threads and the component pad (by using mechanical pressure), and the adhesive is only used for fixing this mechanical contact [14]. UV light curing is possible for an NCA. It can provide a high-speed, low-temperature, inexpensive, and highly automation-friendly contact process for creating smart textiles.

The greatest challenge is the reliability of each connection method in various environments where the product is expected to be used. As mentioned above, wearable textiles have to withstand various conditions that are caused by many factors. One of them is sweat, which might be a significant factor, e.g., for smart textile sports garments. Thus, the reliability testing of joints between SMD chip resistors adhesive bonded or soldered onto textile ribbons during accelerated aging by artificial sweat is important. Moreover, reliability testing of fully printed encapsulated sensors on textile ribbons during sweat aging is also important.

The two basic factors characterizing the environment are temperature and relative humidity. In recent years, graphene-based sensors of these two quantities were developed. The reasons for using graphene are its unique properties, which include high electrical and thermal conductivity, high mechanical resistance and material flexibility, and the possibility of wide modifications to the graphene surface [18,19]. These properties make graphene suitable for use in flexible and wearable sensors [20]. In addition, these sensors can be prepared in a printed form [21].

2. Materials and Methods

For the experiment, three types of samples were used. The first sample type was a ribbon with ten chip resistors connected by soldering and encapsulated by an NCA. The second sample type was a ribbon with ten chip resistors connected by an adhesive bonding technique with an NCA and encapsulated by the same NCA. The third sample type was ribbons with a humidity sensor fully printed by graphene paste through a stencil and partially encapsulated by an NCA. The details of the sample labels can be seen in Table 1.

Before the conducting of our experiment, a few hypotheses were made. The first hypothesis was that the electrical resistance of the bonded or soldered adhesive joint between the SMD component metallization and conductive track woven in the ribbons will be reliable during accelerated aging by artificial sweat. The second hypothesis was that stretching of the ribbon at 150% of its relaxed length during the soldering or adhesive bonding process could slightly improve the sample reliability during aging because the internal tension of the ribbon around the joint is different, and the penetration of sweat into the joint may vary. The third hypothesis was that fully printed and partially encapsulated humidity sensors on the ribbons would be reliable during accelerated aging by artificial sweat.

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Sample type		1—Soldere	d resistors		2—NCA bonded resistors					
Sample marking	1A	1B	1C	1D 2A		2B	2C	2D		
Material used		SnBi sold	er paste		UV-curable acrylic-based adhesive (NCA)					
Stretching	No (100%)	Yes (150%)	No Yes (100%) (150%)		No (100%)	Yes (150%)	No (100%)	Yes (150%)		
Sweat type	Alkaline	Alkaline	Acidic	Acidic	Alkaline	Alkaline	Acidic	Acidic		
Sample type	3—Graphene sensors									
Sample marking	3A	3B	3C	3D	3E	3F	3G	3H		
Material used	G—square 3 × 3		T—square 3 × 3		G—rectangle 1×3		T—rectangle 1×3			
Stretching	No (100%)									
Sweat type	Alkaline	Acidic	Alkaline	Acidic	Alkaline	Acidic	Alkaline	Acidic		

Table 1. Table of used sample types and sample labels.

2.1. Electrically Conductive Textile Ribbons

In the experiment, specially developed electrically conductive textile stretchable ribbons were used. These ribbons are protected by patent No. 308614 "Conductive elastic woven fabric, in particular conductive elastic woven ribbon". These ribbons consist of polyester threads in a warp and weft (strength and durability), rubber threads in a warp (stretchability), and special hybrid conductive threads in a warp (electrical conductivity along the ribbon). The hybrid conductive threads consist of continuous PES monofilaments and eight silver-plated copper conductive microwires for each thread. The ribbon is woven in a stretched state; thus, the ribbon shrinks after its production, and the hybrid yarns inside are naturally arranged in the shape of a horseshoe. This allows the ribbon to be repeatedly stretched by up to 170% of its relaxed length without damage. A schematic cross-section of the ribbon is shown in Figure 1.

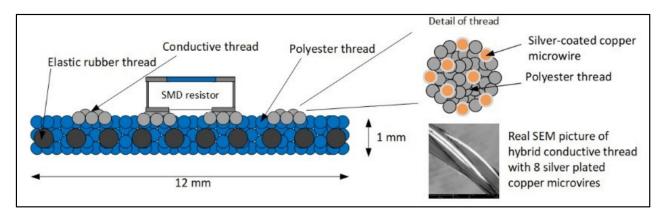


Figure 1. Schematic cross-section of the conductive textile ribbon used in the experiment.

The ribbon has four conductive traces woven inside with a spacing of 2.54 mm, which is the standard spacing for the electronic industry. Each trace has an electrical resistivity in the range of ohms per meter ribbon length and consists of five hybrid conductive threads (i.e., 40 conductive microwires for each trace). This resistance is sufficient for most e-textile applications. The pros of these ribbons are accessibility of conductive traces from the surface and good solderability of the microwires used in the hybrid threads, compared to the silver metallized polyamide threads commonly used in standard conductive ribbons. The traces in the ribbon for the first and second sample types were interrupted every three centimeters by hundreds of amps. Electric discharge was generated by a resistance spot welder from the Sunstone company (see Figure 2) to enable individual electrical resistance measurement of each component by the four-point probe sensing method.

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Figure 2. Details of trace interruption by electric discharge.

2.2. First Sample Type—Soldering (1A, 1B, 1C, 1D)

Ten SMD chip resistors of type 1206 with zero ohm resistance were used for each ribbon for the first sample type. These resistors were soldered perpendicular to the ribbon traces, and the resistors perfectly fit the spacing of the ribbon conductive traces. Low-temperature tin–bismuth solder paste with a melting point of $139\,^{\circ}\text{C}$ was used for soldering chip resistors onto the ribbon traces. The composition of this paste is 40–50% tin, 4.5–6.5% rosin (ROL0), and 50–60% bismuth [22]. The soldered sample production procedure was as follows. The solder paste was applied onto the two central traces by dispensing. After that, the resistor was mounted onto the paste and gently pushed by tweezers into the traces. Then, hand soldering by hot air was carried out. Finally, each soldered resistor was encapsulated by a nonconductive UV curable adhesive to ensure higher chemical resistance and better mechanical properties. One part of the soldered samples was soldered and encapsulated in the stretched state (i.e., the ribbons were stretched to 150% of their relaxed length and fixed in this stretched position for the whole manufacturing process).

2.3. Second Sample Type—Adhesive Bonding (2A, 2B, 2C, 2D)

For the second sample type, the same type and positions of chip resistors as the first sample type were used. An electrically nonconductive, UV-curable, acrylic-based adhesive (NCA) was used for the mechanical connection of chip resistors onto the ribbons. The composition of the adhesive is 30–60% isobornyl acrylate; 10–30% N,N-dimethylacrylamide; 1–10% [3-(2,3-epoxypropoxy)propyl] trimethoxysilane; 1–5% gamma-glycidoxypropyl trimethoxysilane; 1-5% silica, amorphous, fumed, crystal-free; 1-5% diphenyl-2,4,6trimethylbenzoyl phosphine oxide; and 0.1-1% 2-hydroxyethyl acrylate. This adhesive has the ability to fluoresce and a wide operating temperature range $(-40 \, ^{\circ}\text{C} \text{ to } +150 \, ^{\circ}\text{C})$ [23]. The adhesive bonded sample production procedure was as follows. The adhesive was applied onto the ribbon between two central traces by pin transfer. After that, the resistor was mounted to the adhesive and mechanically pushed by a thorn into the conductive traces in the ribbon with a downforce of 30 N. The pressure caused adhesive extrusion from the space between the resistor pads and conductive traces on the ribbon. The adhesive reached a pointwise zero thickness, allowing intimate contact between the metal parts. The resistor pads were also pressed into the conductive traces, and direct mechanical-electrical contact was made; see the X-ray-computed tomography pictures in Figure 3. Then, the adhesive, still under pressure, was cured by UV light for mechanical fixation of the direct contact. The whole principle of this adhesive bonding technique is shown in Figure 4. Finally, each adhesive-bonded resistor was encapsulated by the same adhesive to ensure higher chemical resistance and better mechanical properties. One part of the adhesiveEnergies **2022**, 15, 506 5 of 14

bonded samples was bonded and encapsulated in the stretched state (i.e., the ribbons were stretched to 150% of their relaxed length and fixed in this stretched position for the whole manufacturing process).

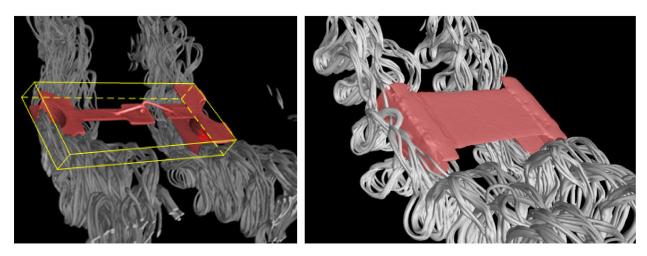


Figure 3. Examples of SMD chip components connected by an NCA onto a conductive ribbon.

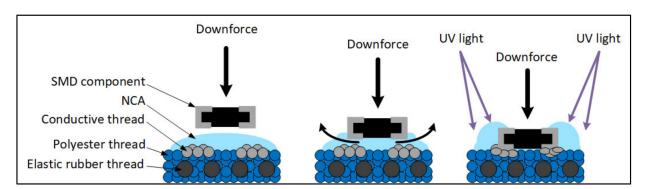


Figure 4. Principle of electrical connection of the SMD resistor to the ribbon by the NCA bonding technique.

2.4. Third Sample Type—Humidity Graphene Sensors (3A, 3B, 3C, 3D, 3E, 3F, 3G, 3H)

For the third sample type, two different graphene-based, screen-printing pastes were used to prepare humidity sensors. Both pastes were purchased from Sun Chemicals and differed in their solid content and sheet resistivity. The first paste, marked G, is type C2131121D3 with a solid content of 41% and a sheet resistance of 10 ohms/sq at a 25 μm layer thickness. The second paste, type C2171023D1, designated T, has a solid content of 35% and a sheet resistance of 7.5 ohms/sq at a layer thickness of 25 μm .

These graphene pastes were applied onto the ribbon between two central traces by stencil printing. The dimensions of the deposited paste were 3 mm \times 3 mm \times 0.12 mm and 1 mm \times 3 mm \times 0.12 mm. After that, the samples were dried in a laboratory drying oven at 60 °C for 60 min. Finally, each sensor was partially encapsulated by the same adhesive to ensure higher chemical resistance and better mechanical properties. After encapsulation, final drying of the sample was conducted in a laboratory drying oven at 60 °C for 60 min.

2.5. Electrical Resistance Measurement

The electrical resistance of each sample was measured by a Keithley 2701 device by the four-wire sensing method of resistance measurement in a defined environment: temperature of 25 $^{\circ}$ C and relative humidity (RH) of approximately 50%.

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2.6. Sweat Aging

In practice, there are two types of human sweat: alkaline and acidic, which depend on the type of human skin. In our experiment, artificial sweat for both sweat types was prepared according to the international standard ISO105-E04 [24]. The compositions of the sweats can be seen in Table 2.

Table 2. Table of artificial sweat composition.

Chemical Composition (Formula)	Alkaline Solution pH = 8.0	Acidic Solution pH = 5.5 Amount		
	Amount			
L-histidine monohydrochloride monohydrate (C ₆ H ₉ O ₂ N ₃ .HCl.H ₂ O)	0.5 g	0.5 g		
Sodium chloride (NaCl)	5 g	5 g		
Disodium hydrogen orthophosphate dodecahydrate (Na ₂ HPO ₄ .12H ₂ O)	5 g	-		
or Disodium hydrogen orthophosphate dihydrate (Na ₂ HPO ₄ .2H ₂ O)	2.5 g	2.2 g		
Total volume	1000 mL	1000 mL		

Accelerated aging was designed and realized in three stages with the same procedures and different time steps; the details are shown in Table 3 and Figure 5.

Table 3. Table of the resulting experimental aging times.

Legend												
30 min Swea	24 h Sweat					7 Days Sweat						
Cycles	1	2	3	4	5	6	7	8	9	10	11	12
Overall hours in sweat	0.5	1	1.5	2	2.5	3	3.5	4	4.5	5	5.5	6
Overall hours in water	0.5	1	1.5	2	2.5	3	3.5	4	4.5	5	5.5	6
Days in liquid	0.04	0.08	0.13	0.17	0.21	0.25	0.29	0.33	0.38	0.42	0.46	0.5
Cycles	13	14	15	16	17	18	19	20	21	22	23	24
Overall hours in sweat	6.5	7	31	55	79	103	127	151	319	487	655	823
Overall hours in water	6.5	7	7.5	8	8.5	9	9.5	10	11	11	12	12
Days in liquid	0.54	0.58	1.6	2.6	3.6	4.7	5.7	6.7	13.7	20.8	27.8	34.8

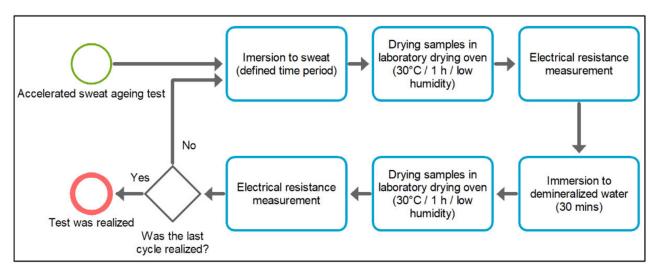


Figure 5. Procedure of conducting accelerated aging by artificial sweat test.

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3. Results and Discussion

3.1. Conductive Joint Samples

The measured values were statistically analyzed, and boxplot diagrams were prepared (see Figures 6 and 7). The results show that the electrical resistance of joints for soldered samples during aging by sweat is low and stable for both sweat types without significant differences between sample types. The results of the adhesive-bonded samples show that the electrical resistance grows during aging and that the deviation of samples increases. In the diagrams, small differences in the boxplots of new adhesive-bonded samples are observed. Our manufacturing is an experimental prototype production conducted by hand (not automated), and the placement of the thorn onto the component can be slightly different for each sample, which can lead to the production of a lower-quality sample. Such a sample can be degraded faster, which causes greater differences after aging. This problem can be eliminated by preparing a large number of samples, but this was not possible because the preparation of samples is expensive and time-consuming. We solved this problem by preparing a line chart with the median of values (see Figure 8). In the case of median values, the outliers are suppressed. This chart shows that adhesive-bonded samples have a higher electrical resistance after their production and that the electrical resistance of adhesive-bonded samples grows during the first few hours in sweat; however, after this initial growth, the values are relatively stable, with only a slight increase over time. The results also show that samples under alkaline sweat have a slightly worse resistance, but this difference is relatively insignificant.

The samples after the whole aging process (i.e., 823 h in sweat) were submitted for microscopic observation to find visual changes (see Figures 9 and 10). The degradation of the sample is clearly visible; however, encapsulation clearly fulfilled its purpose, and the joint under encapsulation was protected against humidity and sweat. As can be seen, the residues are not verdigris from copper wires, but only pigments transferred from blue polyester threads to the transparent polyester threads inside the hybrid conductive threads.

Metallographic cross-sections of ribbon traces after artificial sweat aging were prepared and observed by an Olympus LEXT OLS 5000 confocal laser scanning microscope (see Figure 11). The observation shows that transparent polyester threads were slightly colored. The details of electrically conductive microwires show that some of these wires are degraded under the silver coating, but our experiment was much more challenging than the standard usage of smart textiles.

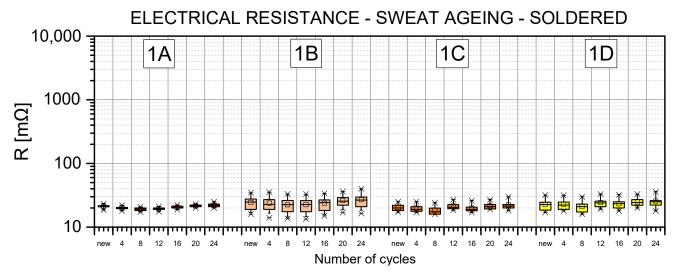


Figure 6. Boxplot diagram of soldered joint (first sample type) electrical resistance for samples 1A, 1B, 1C and 1D.

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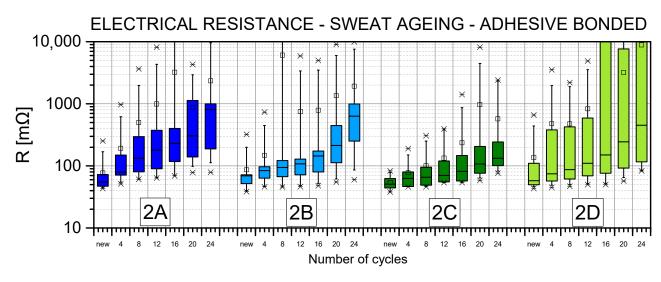


Figure 7. Boxplot diagram of adhesive-bonded joint (second sample type) electrical resistance for samples 2A, 2B, 2C and 2D.

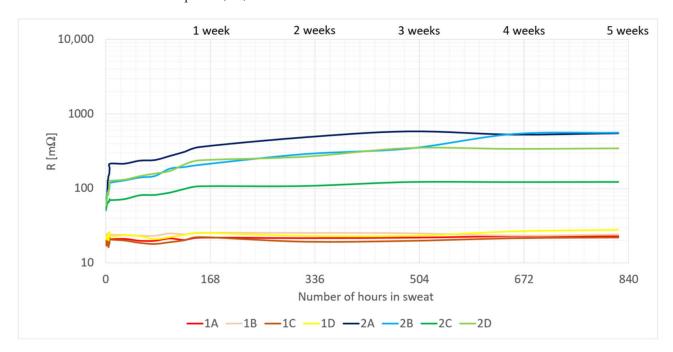


Figure 8. Medians of electrical resistance for the first (1A, 1B, 1C, 1D) and second (2A, 2B, 2C, 2D) sample types tested under sweat.



Figure 9. Example of a soldered sample after aging by artificial sweat.

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Figure 10. Details of the sample after aging by artificial sweat.

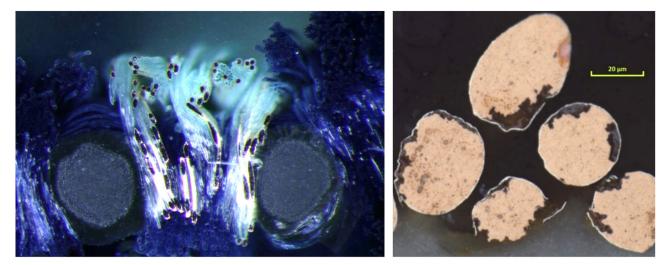


Figure 11. Microscopic observation of metallographic cross-sections of ribbon traces after artificial sweat aging.

3.2. Humidity Sensor Samples

Graphene occurs in the paste in the form of small multilayer flakes. These flakes contain defects in their structure (mainly at the edges), where oxygen groups such as carboxyl, hydroxyl and epoxy groups are most often attached. There are two mechanisms that can explain the moisture dependence of graphene-based materials. (i) Water molecules easily bind to the oxygen-side groups in the graphene structure via physisorption forces. Graphene essentially has a p-type conductivity, where the major charge carriers are holes. Water is an electron donor, and when it interacts with graphene, electrons pass into the graphene structure where they recombine with the holes, thus reducing their concentration. This leads to an increase in the resistance of graphene to moisture during exposure. (ii) Water is a small molecule and, especially at higher humidity levels, can intercalate between individual graphene layers in a single flake or between whole flakes. This causes an increase in the distance between the individual layers in a graphene flake or an increase in the distance between flakes, thus increasing the resistance of the entire layer [25]. These two phenomena can have a synergistic effect at high humidity values.

Humidity tests of graphene-based sensors were performed in a Vötsch VCV $_3$ 7060-5 climatic chamber. The prepared sensors were exposed to a gradual change in humidity from 20% RH to 98% RH (maximum limit of the climatic chamber) with a step of 10% RH, and then back to 20% RH with the same step. The time for stabilization at every level of RH was 90 min. All humidity tests were carried out at a constant temperature of 25 $^{\circ}$ C.

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The response of the sensor to RH is shown in Figure 12. The maximum sensor response at 98% RH is very high, at 98%. The response value is the relative change in sensor resistance calculated according to the following Equation (1):

Response =
$$\Delta R/R0 = ((R - R0)/R0) \times 100\%$$
 (1)

where R (Ω) is the sensor resistance during humidity exposure, and R0 (Ω) is the sensor resistance at the minimum value of humidity (T = 25 °C; RH = 20%). The greater response at a high RH is probably determined by the synergy of the graphene-sensing mechanisms described above. Although there is a hysteresis between the values with an increasing and decreasing humidity, the sensor returns to the original resistance value. As the humidity decreases, the resistance value of the sensor does not return to its original value (value with rising RH). This phenomenon is caused by the insufficient spontaneous desorption of water molecules from the active layer of the sensor. Especially between the individual graphene layers, water molecules are bound by a large binding energy. With a lower content of water molecules in the ambient air, spontaneous desorption improves, and the result is the same resistance value at 20% RH. A very good repeatability of the sensor is shown in Figure 13.

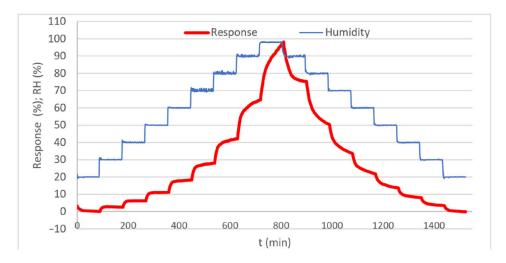


Figure 12. Response of the graphene-based, rectangle-type sensor to RH (condition: temperature of 25 °C).

The prepared sensors were submitted to the same sweat test described above. The measured values were statistically analyzed, and boxplot diagrams were prepared (see Figures 14 and 15). The results show that the electrical resistance of square-type samples during aging by sweat is sufficiently stable for both sweat types. The outliers are relatively significant, but the preparation of a calibration curve for each sample makes the sample applicable, or these samples could be removed. The electrical resistance of rectangle-type samples during aging by sweat is also stable for both sweat types, with a greater variance of values, but the resistance of these samples is much higher. The changes in the resistance during testing in the chamber are also more dramatic and cause a higher sensor sensitivity. Additionally, a line chart with the median values was prepared (see Figure 16). In the case of median values, the outliers are suppressed. This chart shows that rectangle-type samples have a higher electrical resistance, but overall, the resistance of all samples is relatively stable with only a slight increase with time during sweat aging. The results also show that there is no significant difference between samples tested in alkaline and acidic sweat.

The samples after the whole aging process (i.e., 823 h in sweat) were submitted for observations to find visual changes (see Figure 17). The degradation of conductive lines is clearly visible, with a higher intensity for samples aged in acidic sweat. However, encapsulation clearly fulfilled its purpose, and the joint under encapsulation was protected against sweat.

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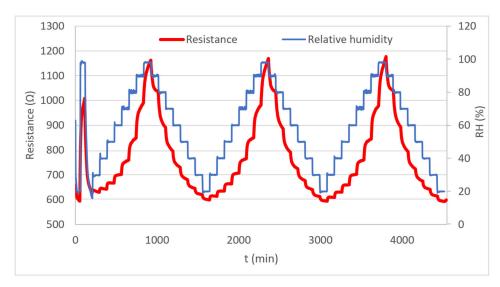


Figure 13. Repeatability of the graphene-based, rectangle-type sensor exposed to three identical humidity cycles.

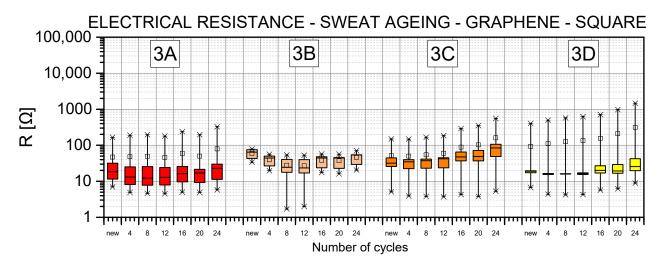


Figure 14. Boxplot diagram of graphene sensor (third sample type—square shape) electrical resistance for samples 3A, 3B, 3C and 3D.

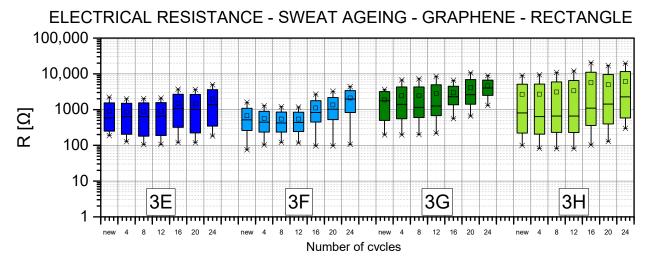


Figure 15. Boxplot diagram of graphene sensor (third sample type—rectangle shape) electrical resistance for samples 3E, 3F, 3G and 3H.

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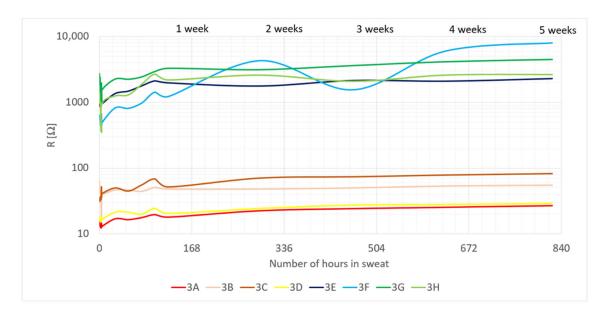


Figure 16. Medians of electrical resistance for the third sample type tested under sweat for samples 3A, 3B, 3C, 3D, 3E, 3F, 3G and 3H.

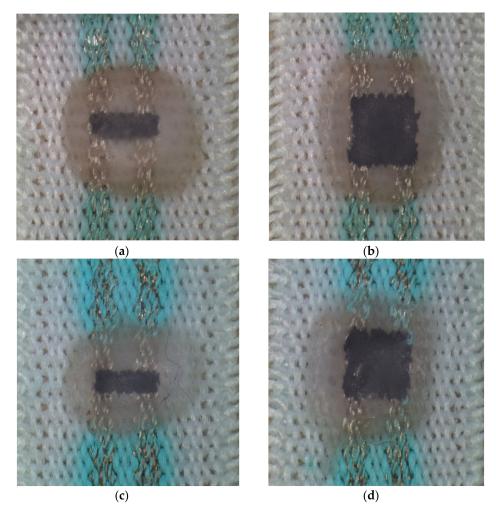


Figure 17. Example of graphene sensor samples after aging by artificial sweat (alkaline—(a,b); acidic—(c,d)).

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4. Conclusions

The results of the electrical resistance of adhesive-bonded or soldered joints of SMD components on conductive textile ribbons with hybrid threads after aging by artificial sweat show that both of these methods are applicable. The first hypothesis was that the electrical resistance of the adhesive-bonded or soldered joint, between the SMD component metallization and conductive track woven in the ribbons, will be reliable during accelerated aging by artificial sweat. This was fully confirmed for soldered samples and partially confirmed for glued samples. The second hypothesis was that the stretching of the ribbon at 150% of its relaxed length during the soldering or adhesive-bonding process could improve the sample reliability during aging. This was not confirmed because both sample types had similar results without significant changes. We suppose that prestretching the ribbon would have an effect on stretching tests or tests that combine stretching with sweat (i.e., more realistic conditions). The soldered samples have lower and more stable results during the whole-tested aging by both types of artificial sweat and can be fully recommended from this view. The adhesive-bonded samples have worse and slightly increasing results, but the values are still usable for some applications (e.g., passive safety illumination by SMD LED components). Both tested technologies have some pros and cons, and both technologies are usable for dedicated applications. Soldering is faster, the electrical resistance is lower, and the contact is more stable; however, soldered joints are harder and more brittle than adhesive-bonded joints. Additionally, the high thermal load during soldering can disrupt the ribbon. Adhesive bonding is slightly slower, the electrical resistance is higher, and the results are worse; however, adhesive-bonded joints are more flexible and can be gently applied to the ribbon at room temperature. Adhesive bonding can also be more suitable for components with multiple leads because conductive bridges between leads are excluded.

The results of fully printed graphene humidity sensors on conductive textile ribbons with hybrid threads show that the realization of this sensor type is possible. The results also show that the rate of humidity change measured by these sensors is in the high tens of minutes. This phenomenon is likely caused by the partial encapsulation of samples, but the intended use of these samples is to detect changes in the moisture of the human body. In this case, the temperature is relatively stable, at approximately $35–37\,^{\circ}$ C, and the change in humidity does not need to be immediately detected. Thus, the measured response characteristics are sufficient for this purpose. The third hypothesis, that fully printed and partially encapsulated humidity sensors on the ribbons will be reliable during accelerated aging by artificial sweat, was confirmed.

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